

PACS-CS における物質・生命科学 I: ナノ形状の量子論

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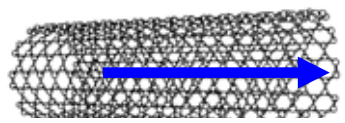
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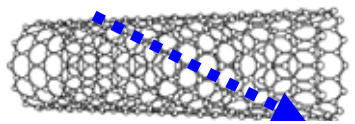
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[Okada & Oshiyama: PRL 91, 216801 (2003), and unpublished results.]
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 - *Higher electron temperature that simulates electron-excitation-induced materials formation*
- What we are planning to do on PACS-CS

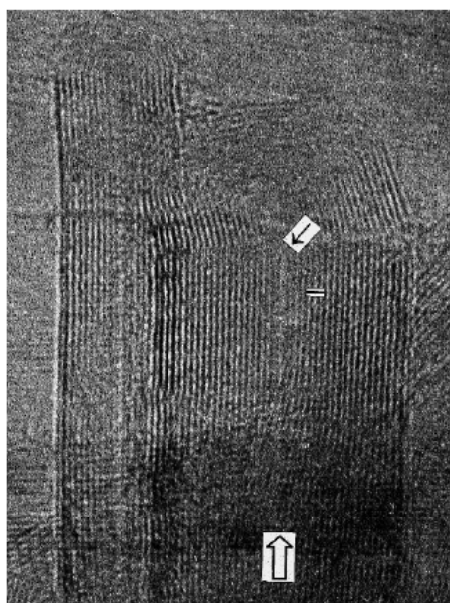
Thin Nanotube in Multiwalled Nanotubes



metallic armchair nt (n,n)



semiconducting zigzag nt (n,0)



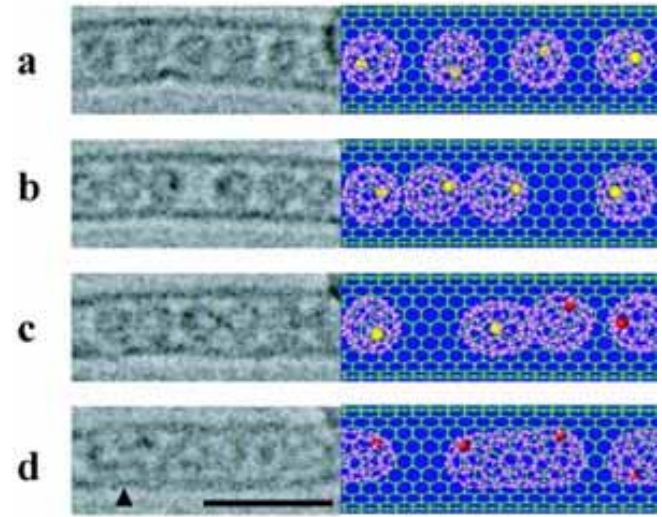
(7,0)@MWNT

L. F. Sun et al.,
Nature 403 384 (2000)

L.-C. Qin et al.,
Nature 408 50 (2000)



4Å-nanotube@MWNT
(3,3), (4,2), (5,0)

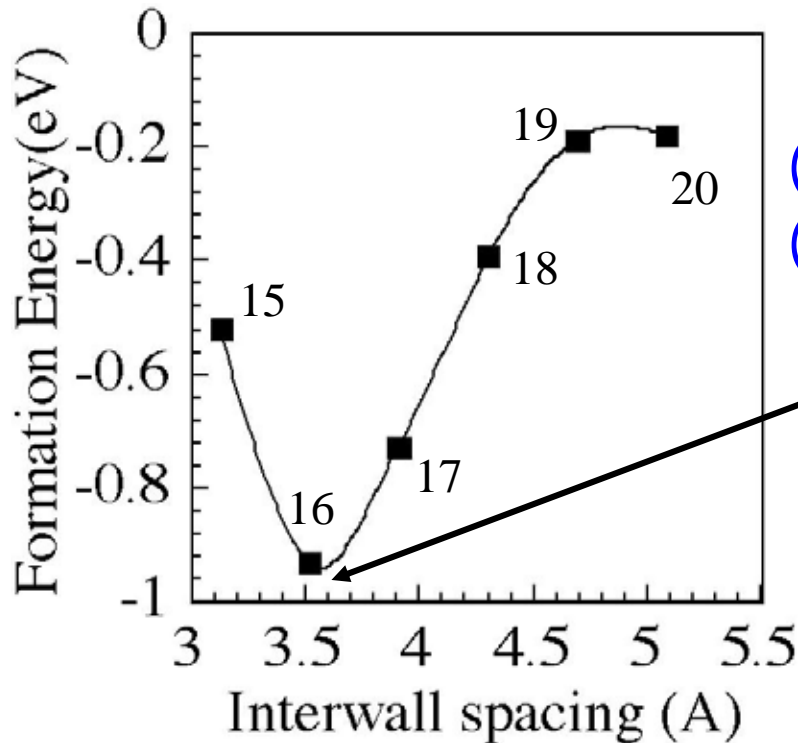


[Toshiya Okazaki (AI ST, Japan):
Sm@C82@CNT]

Electron beam makes
peapods DWNTs

What do we expect for
DWNTs?

Energetics of $(7,0)@(n,0)$



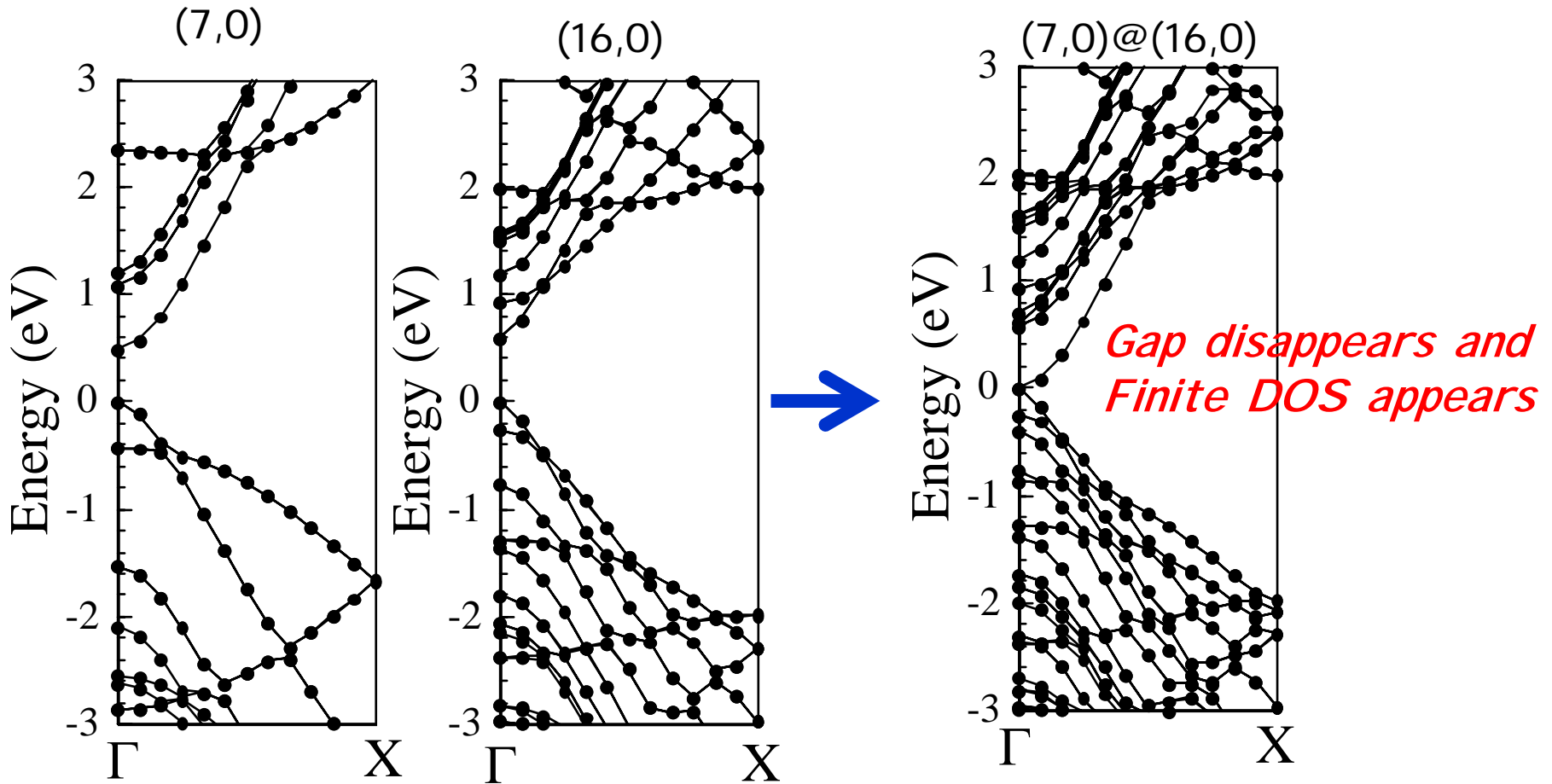
n in $(7,0)@(n,0)$

$(7,0)@(16,0)$ is most stable
 $(7,0)@(17,0)$ is also preferable

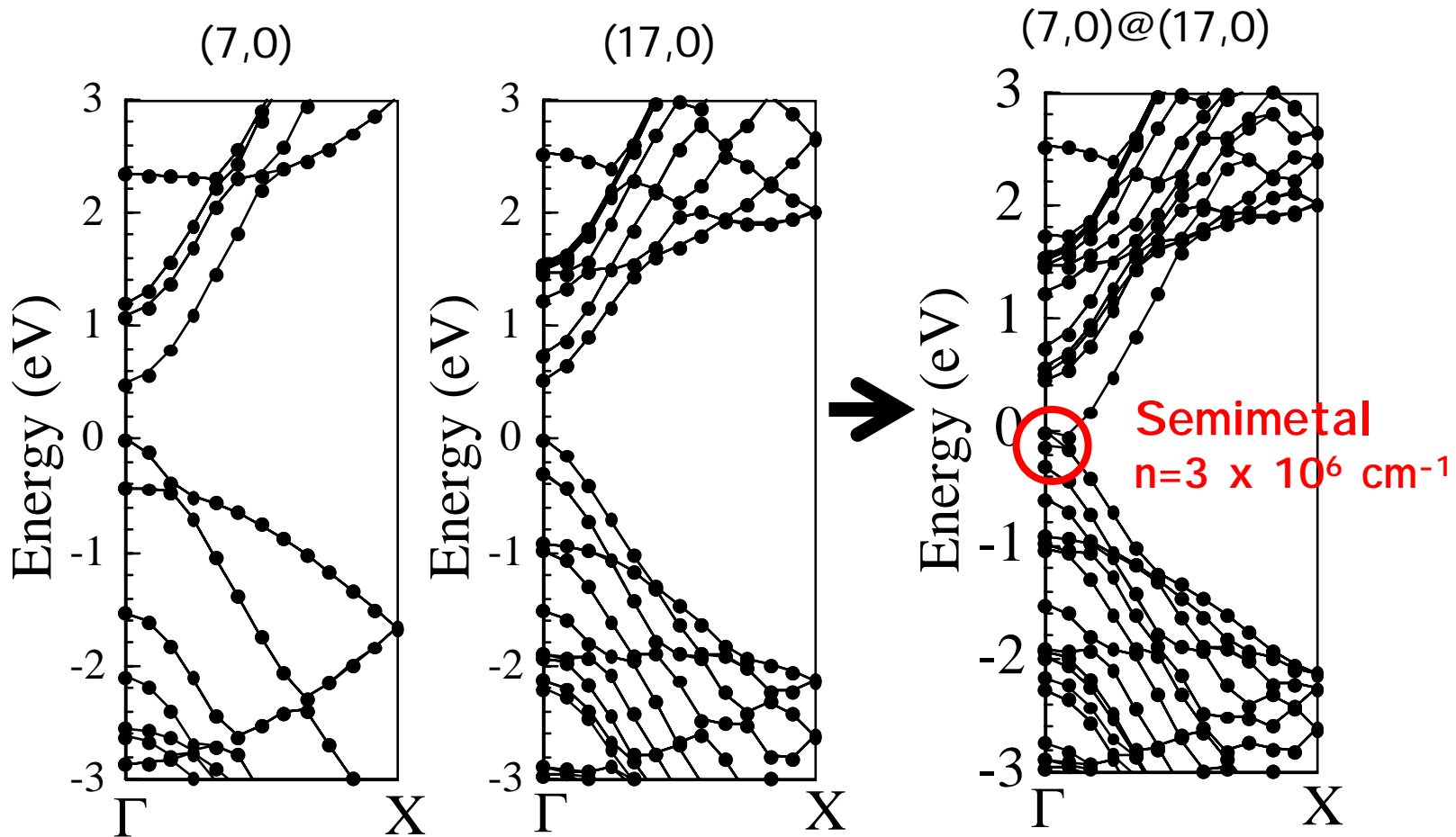
*Spacing is larger than
interlayer distance in
graphite*

Consistent with Electron
Diffraction measurement
by Hirahara & Iijima

Electronic Structure of $(7,0)@(16,0)$

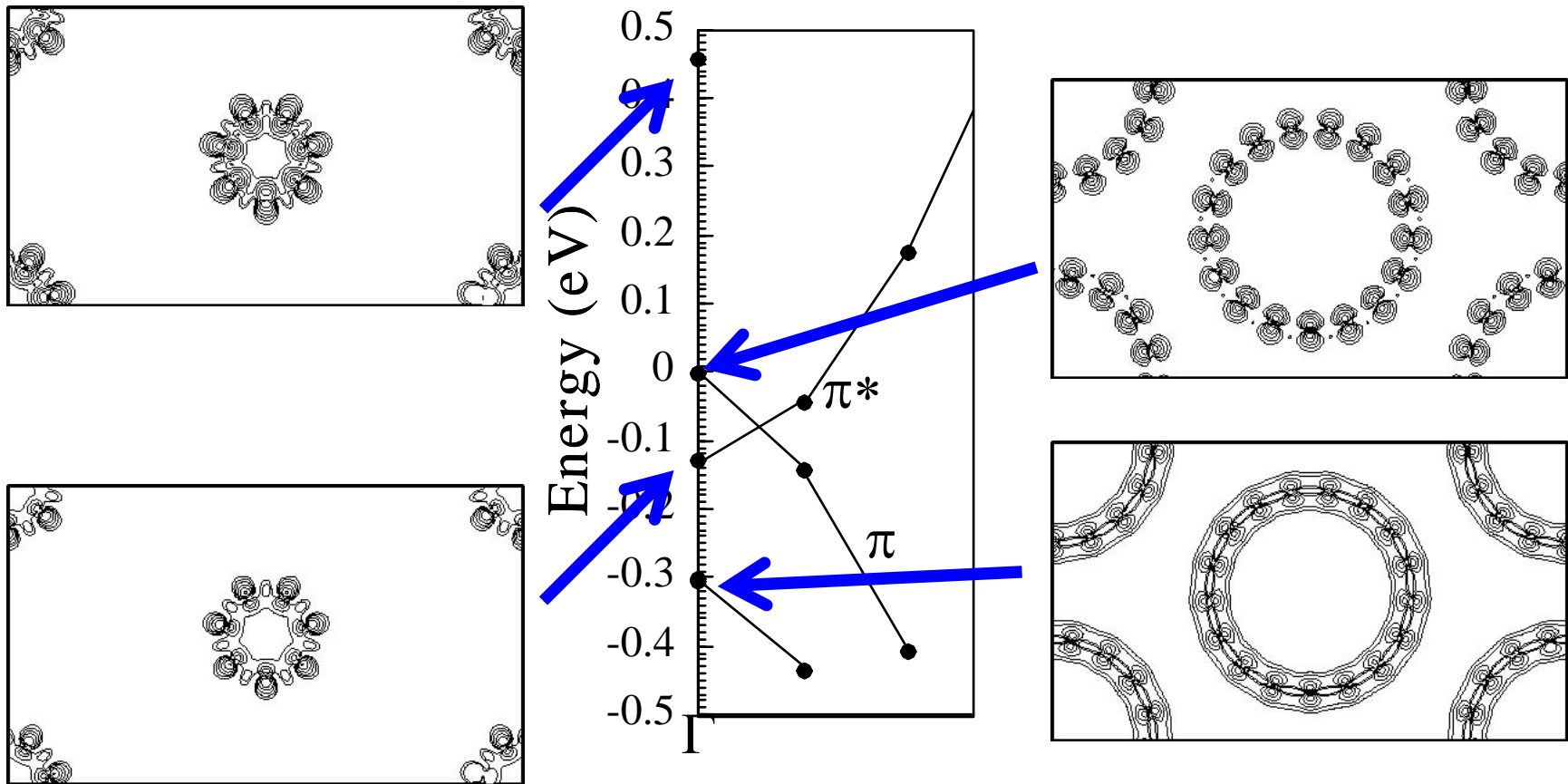


Electronic Structure of (7,0)@(17,0)



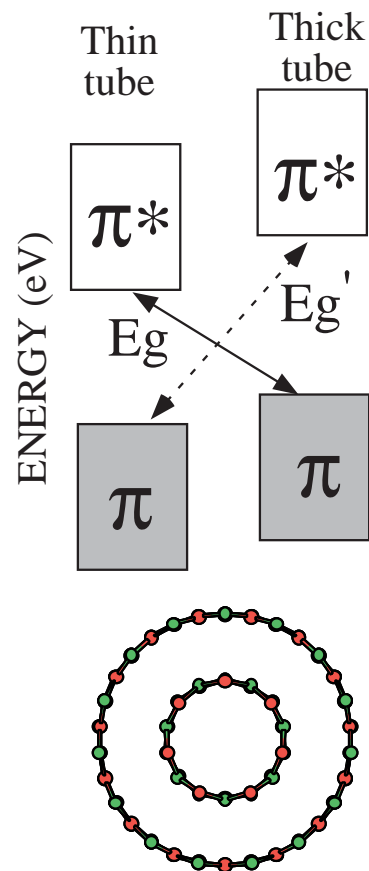
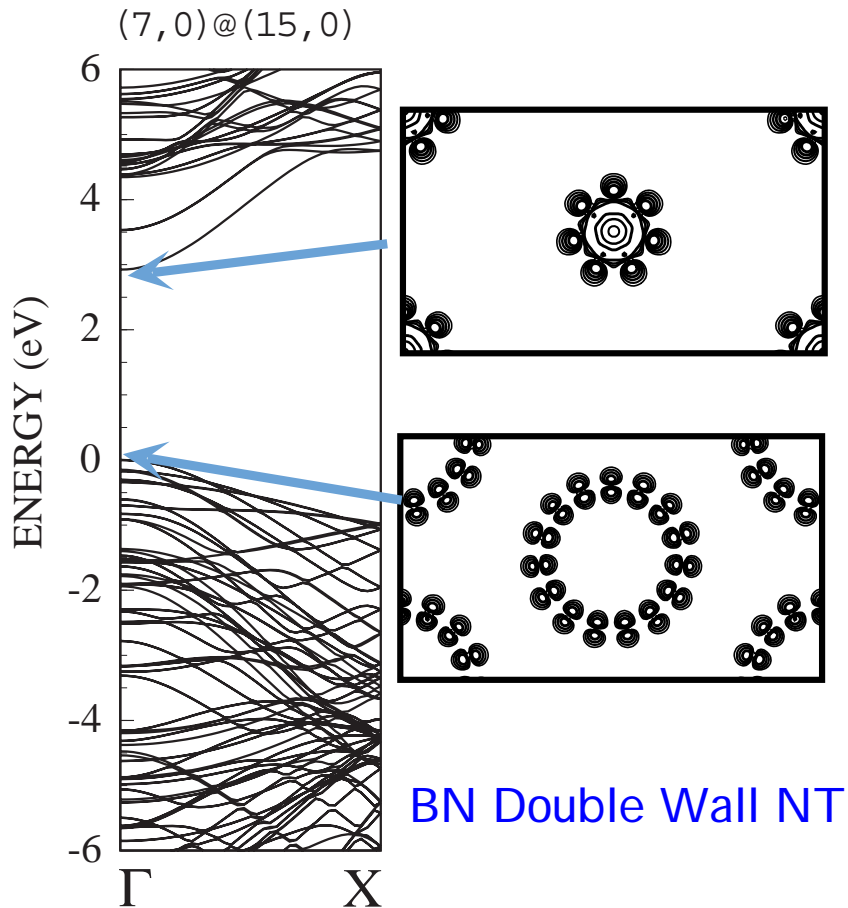
Curvature Induces s-p mixing and It depends on radii

Extension of energy bands near E_F in (7,0)@(17,0)

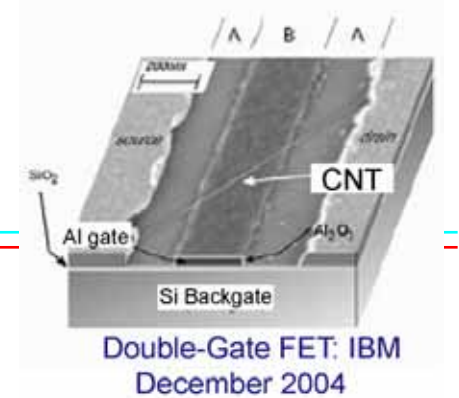


Common to Tubular Structure

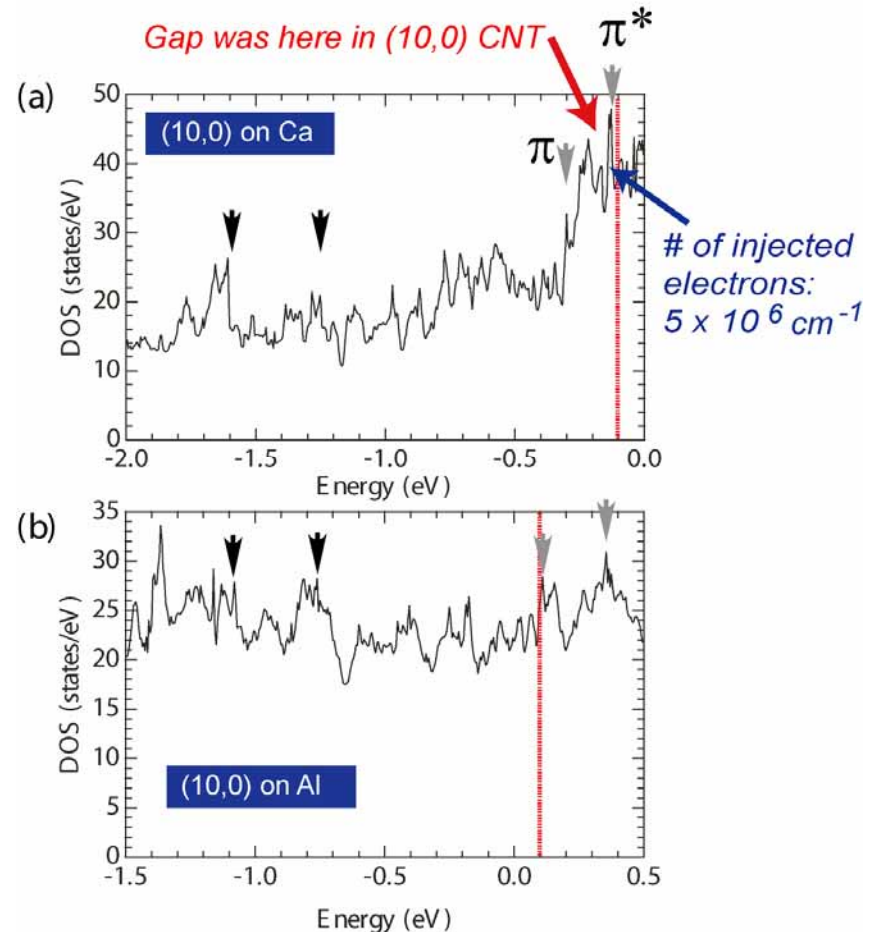
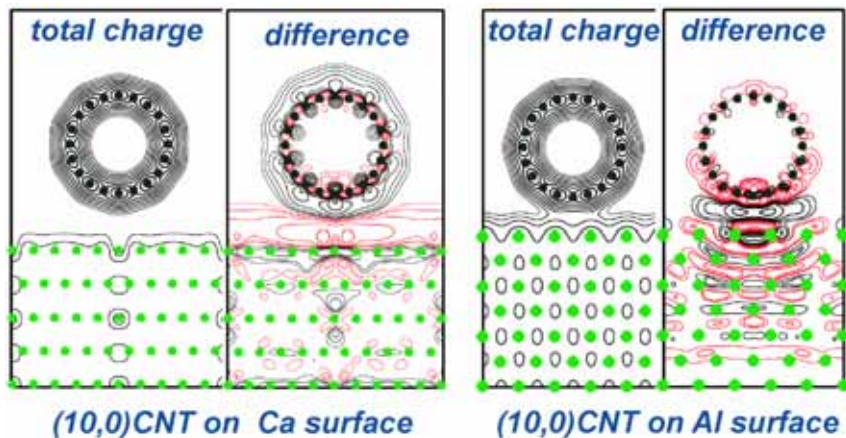
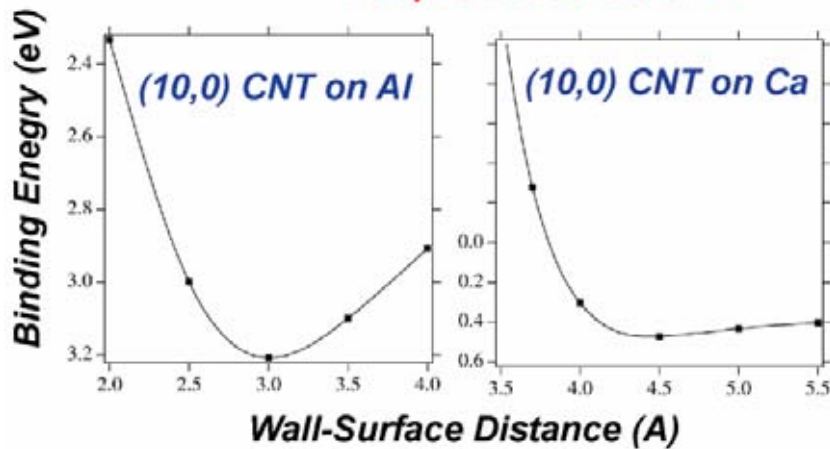
S. Okada, S. Saito & A. Oshiyama: PRB 65, 165410 (2002)



Nanotube/Metal Interfaces: a role of hybridization



Yes, It Does Stick!



Curvature could control number of injected electrons

Thinner (9,0) CNT on Al surface

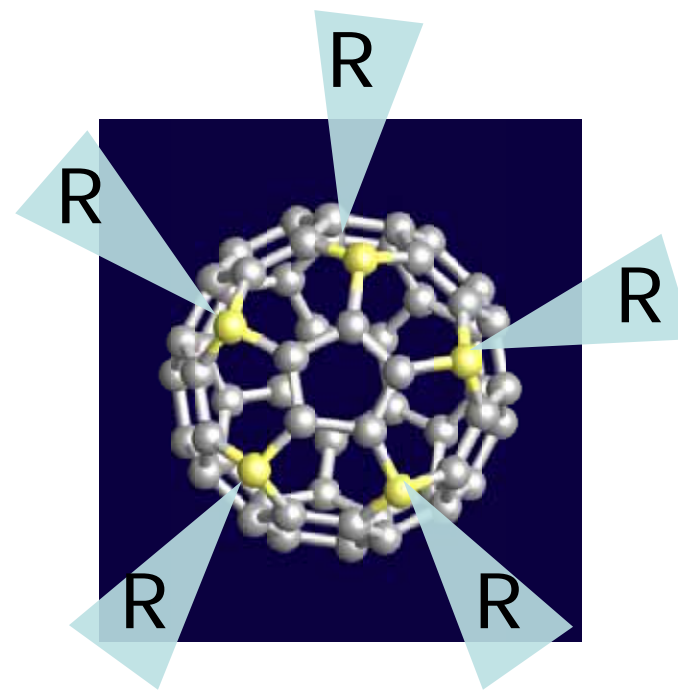
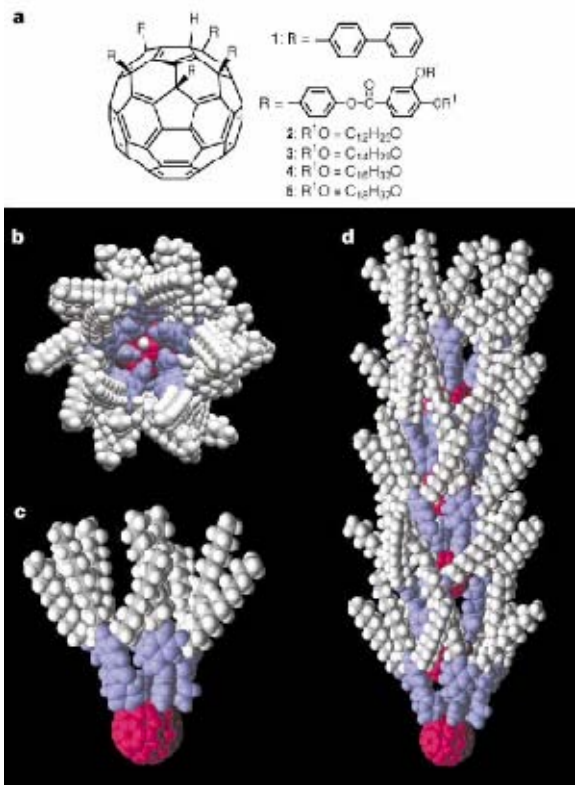
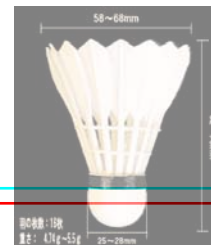
Fermi Level is above original π^ due to small curvature, or in other words, to larger s mixing, and then*

$1.1 \times 10^6 \text{ cm}^{-1}$ electrons are injected from Al to CNT

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Nano-scale badminton Shuttlecock



Chemical modifications

R = biphenyl (C₁₂H₁₀)
 methyl (CH₃)
 (C₆H₅)-Me
 (C₆H₅)-polyacetylene, etc...

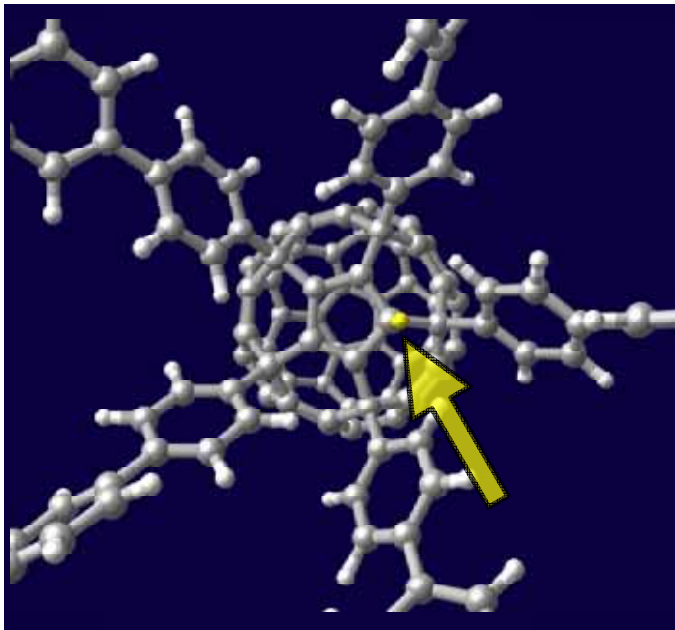
Figure 1 Fullerene derivatives with covalently attached structures. **a**, Chemical formulae of **1** **b**, Top view of **2**. Colored scale (red, yellow, blue) indicates the molecular weight groups, and grey filled chains. **c**, Side view of **2**. **d**, A 3D view of the molecules of **2** based on molecular models calculated based on the XRD data presented in the text.

M. Sawamura et al., Nature, 419, 702 (2002)

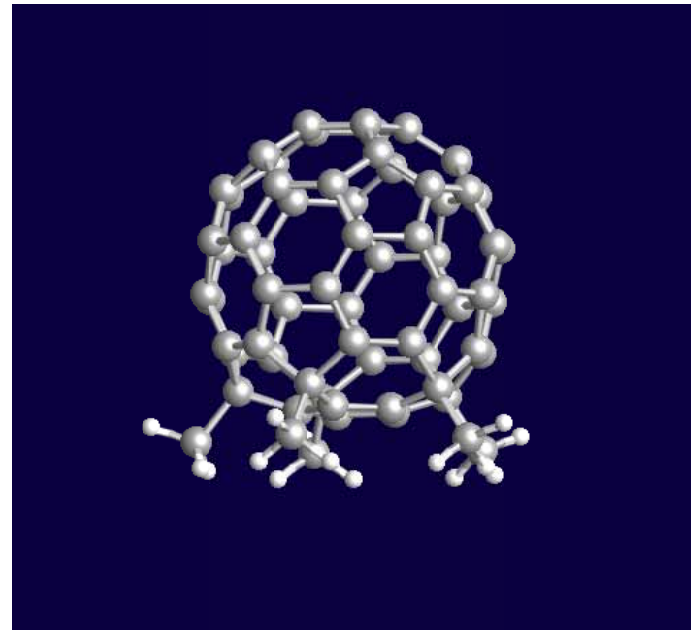
Let's calculate following Nano-scale Shuttlecock

$C_{60}H-(biphenyl)_5$

$C_{60}(biphenyl)_5$

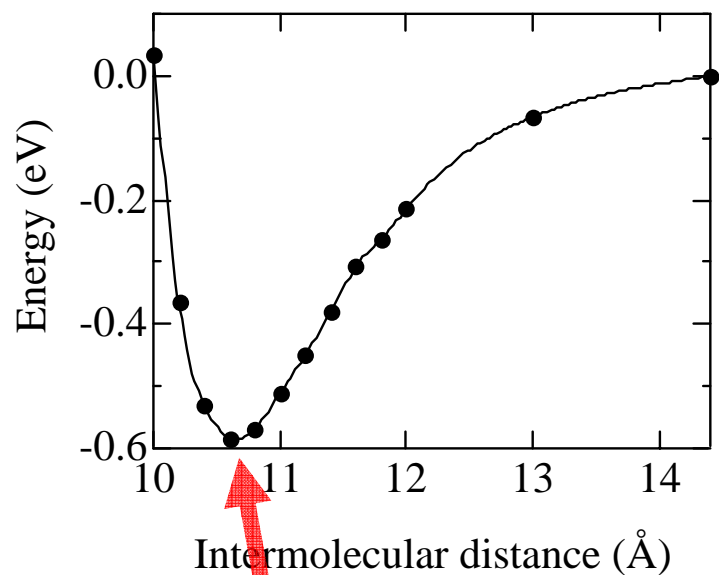
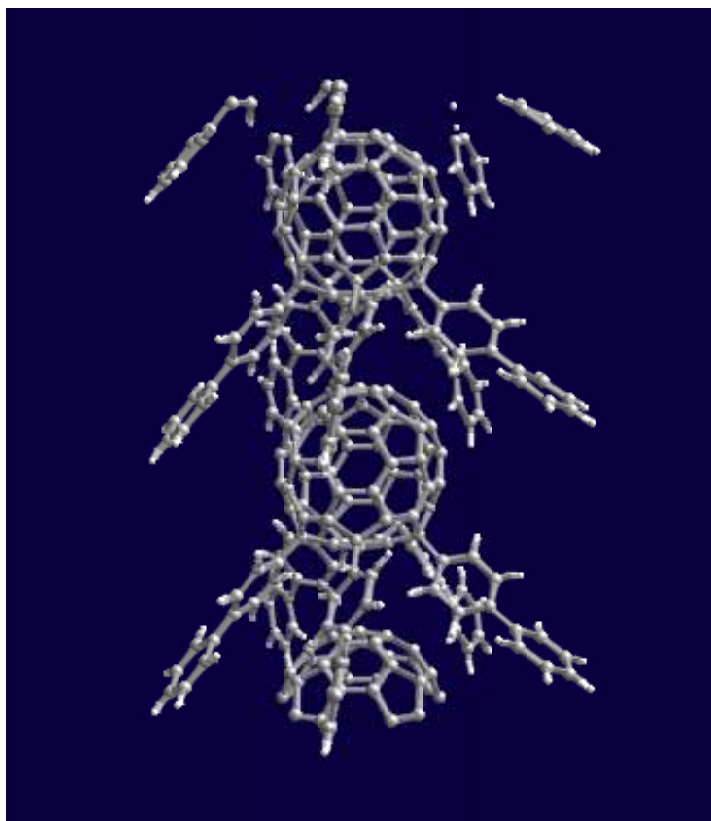


$C_{60}(Methyl)_5$



They are all synthesized

Geometric Structure of C₆₀H-(biphenyl)₅

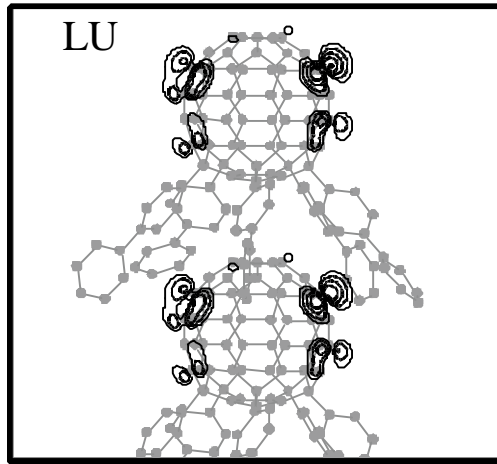


$C_{60}-C_{60} = 10.66 \text{ \AA}$
cf. 10.0 \AA for fcc C_{60}

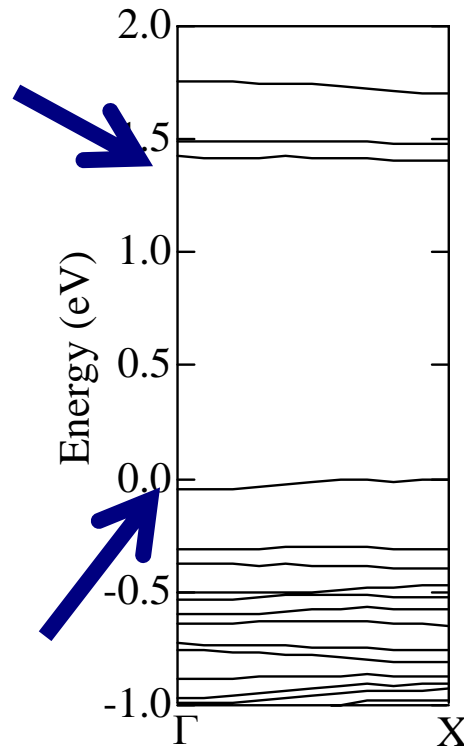
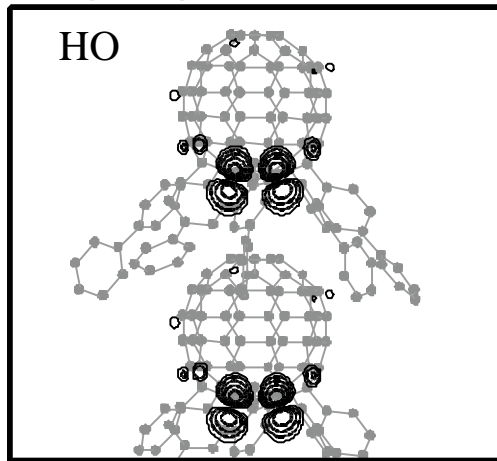
Geometric hindrance (H atom & C₆₀) results in the large intermolecular spacing.

Electronic Structures: C₆₀H-(biphenyl)₅

LUMO



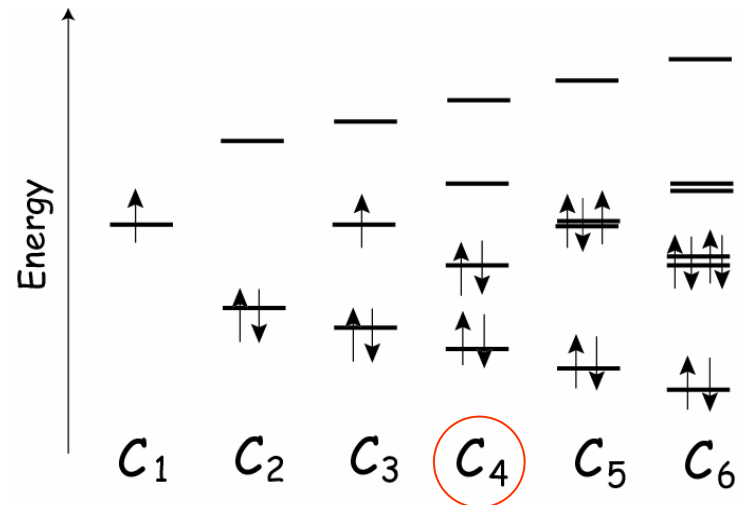
HOMO



➤ Energy gap: 1.4 eV

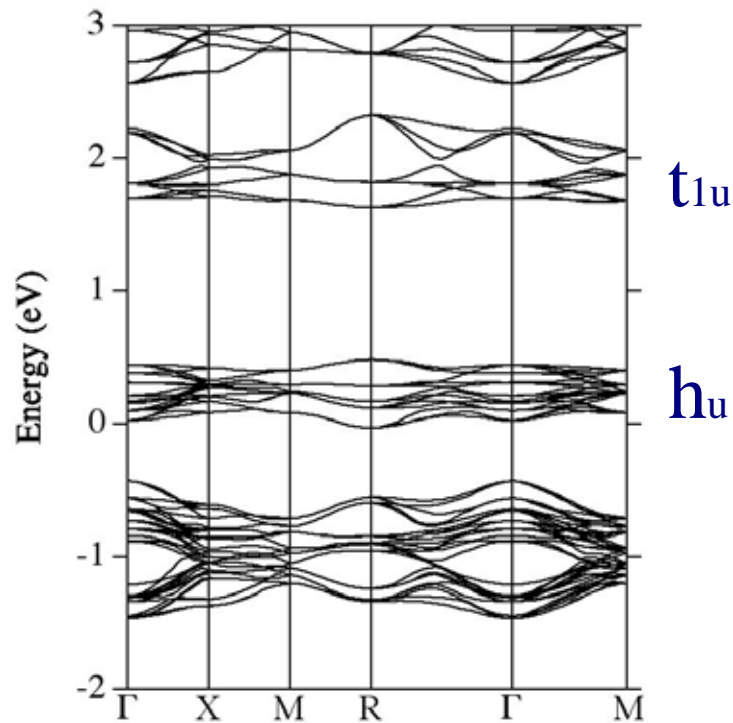
➤ LUMO Band width: 0.05 eV

➤ HOMO Band width: 0.02 eV

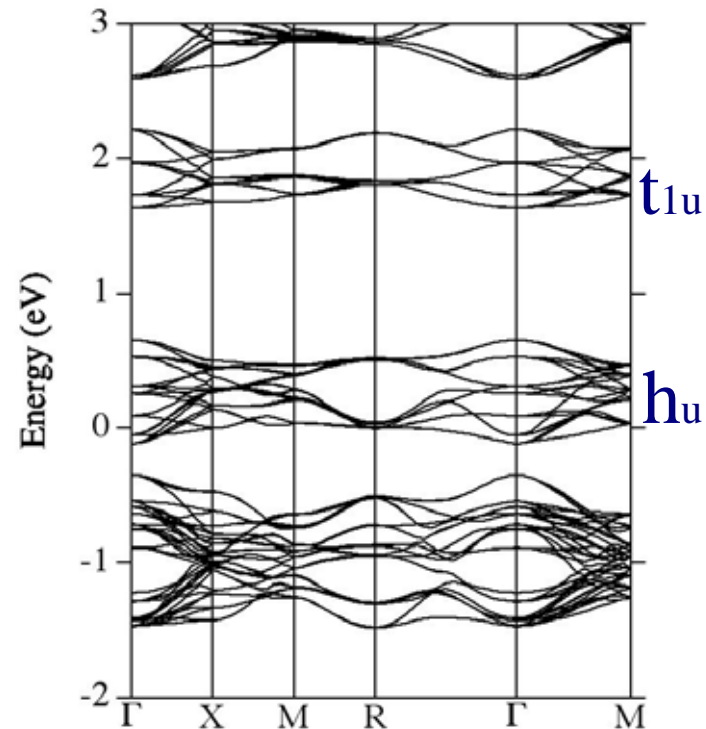


Compare: Electronic Energy Bands of Solid C_{60}

SC phase

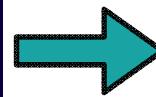
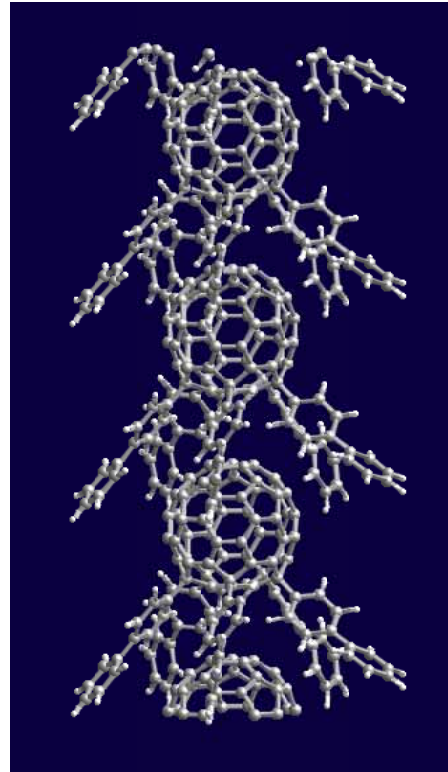
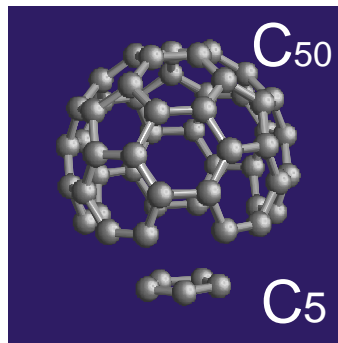
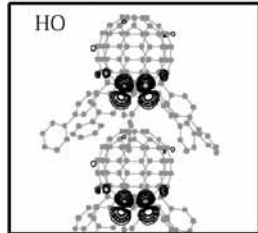
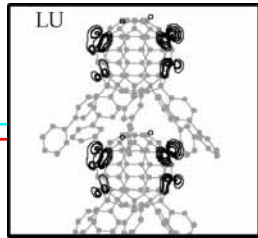


FCC phase

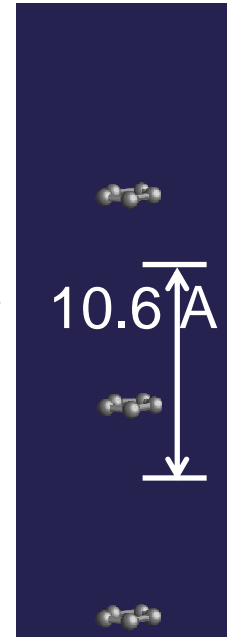


*Energy Gap is a half eV, much larger than 20-50 meV
 C_{60} - C_{60} distance: 10.0(fcc) vs 10.6(shuttlecock)*

New Electron Network

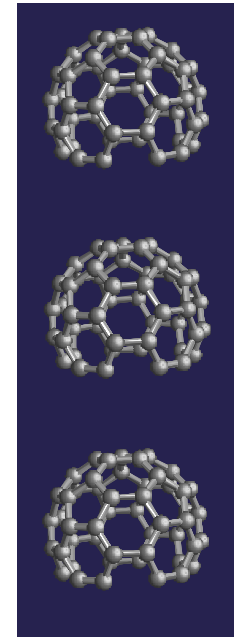


HOMO



C₅

LUMO

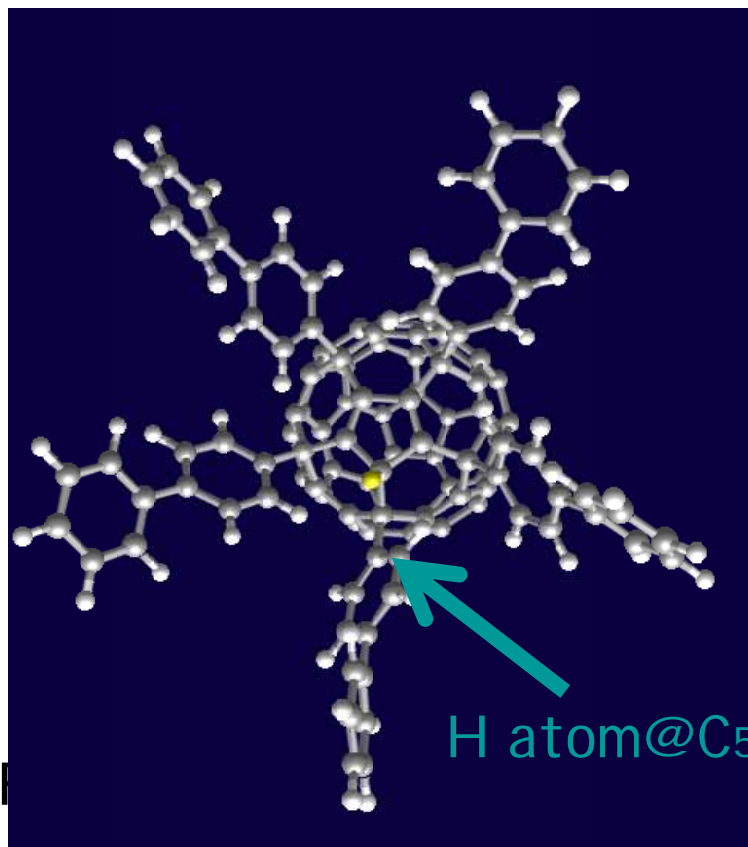


C₅₀

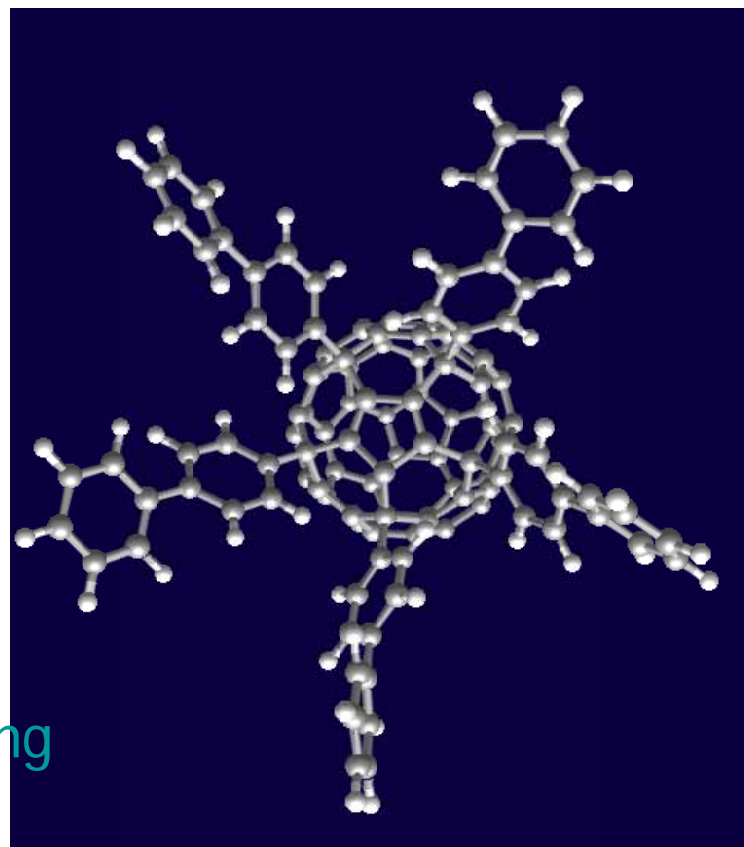
➤ Shuttlecock chain is a new electron system in which attached molecules divide the C₆₀ into C₅ and C₅₀ units.

Hydrogen Removal could Modify Electron States of Shuttlecock

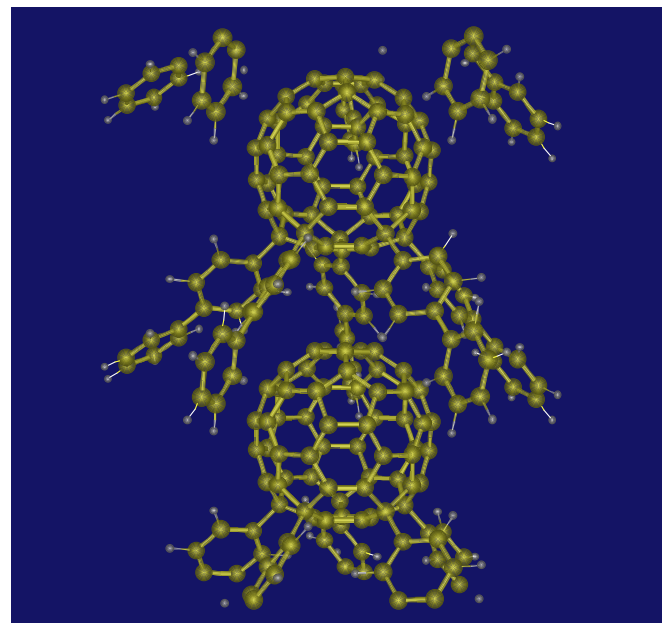
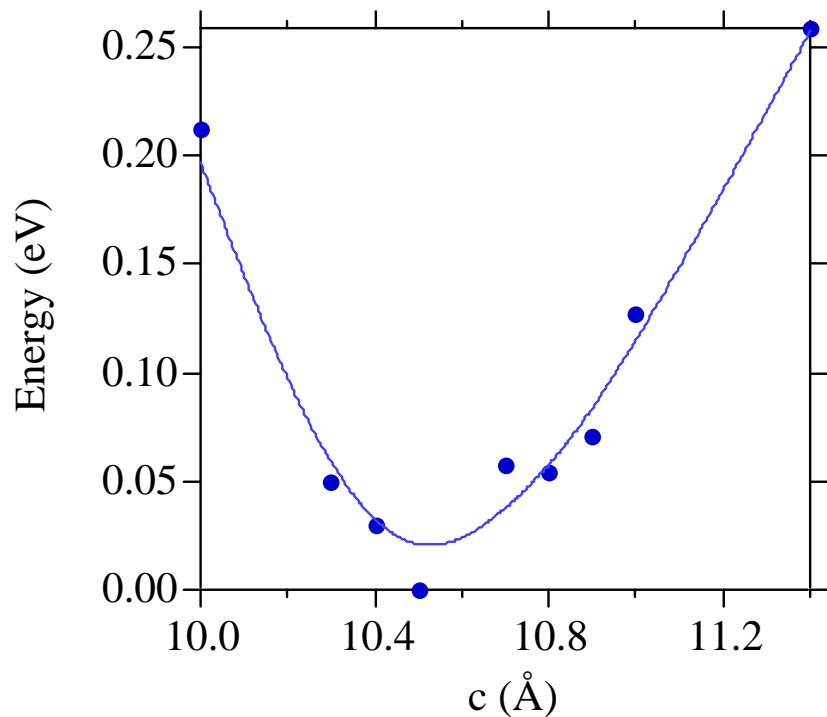
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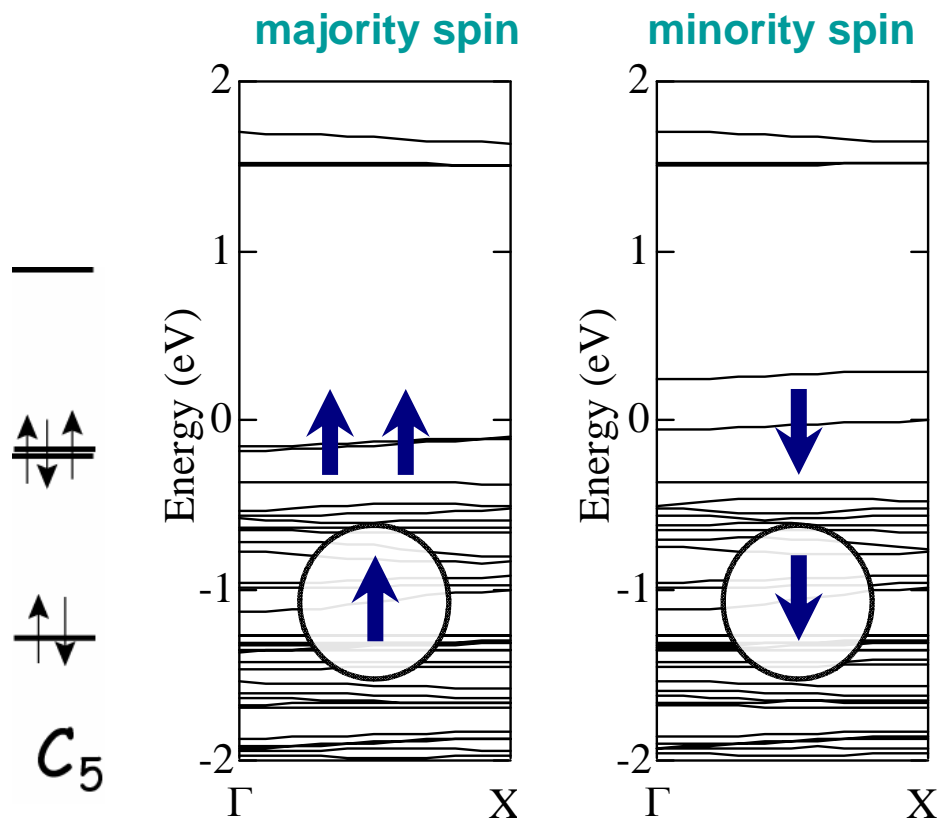
$C_{60}-(biphenyl)_5$



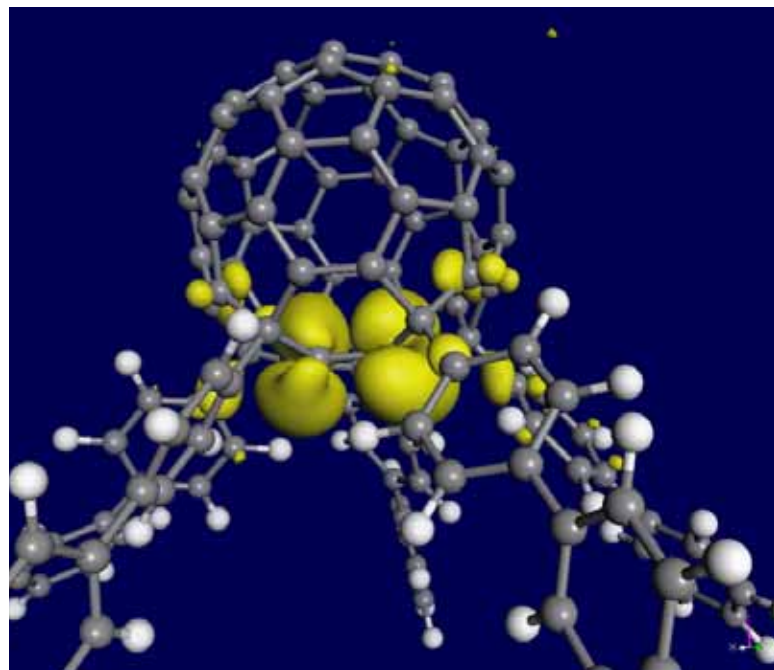
Geometric Structure of C_{60} -(biphenyl)₅



Electronic Structure: C60(biphenyl)5

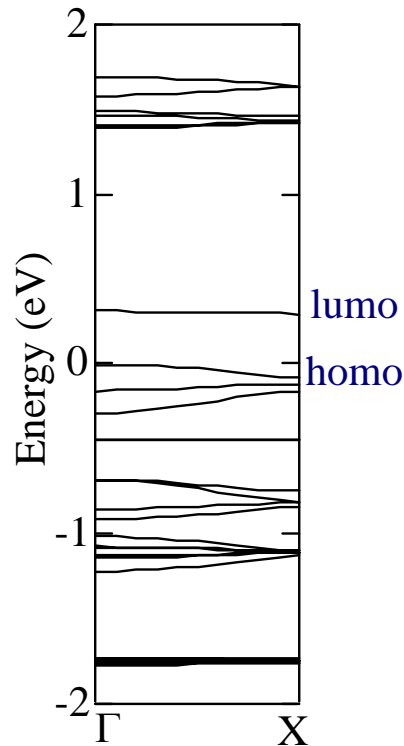


spin: $S - S = 1 / 2$



Spin polarization: $E_{\text{high spin}} = E_{\text{no spin}} - 98 \text{ meV}$

Decoration of Fullerene Makes it Magnetic Shuttlecock



Total Energy (meV/C60-Me5)

Ferro

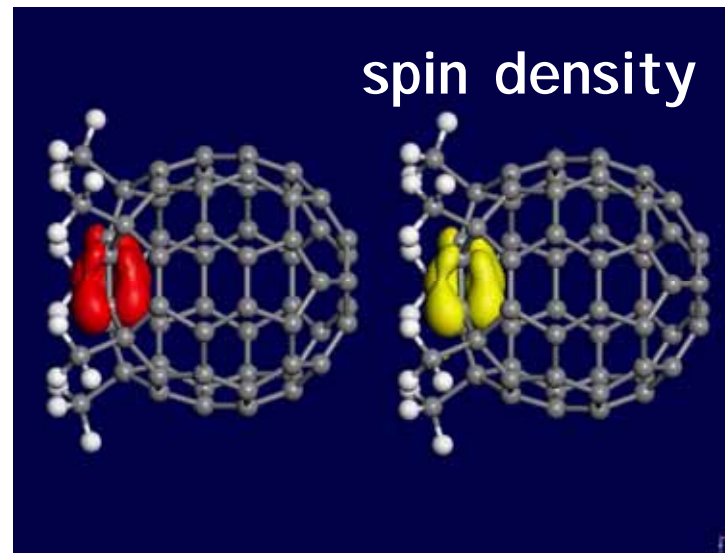
-48

AntiFerro

-58

No spin

0



Antiferromagnetic order!!

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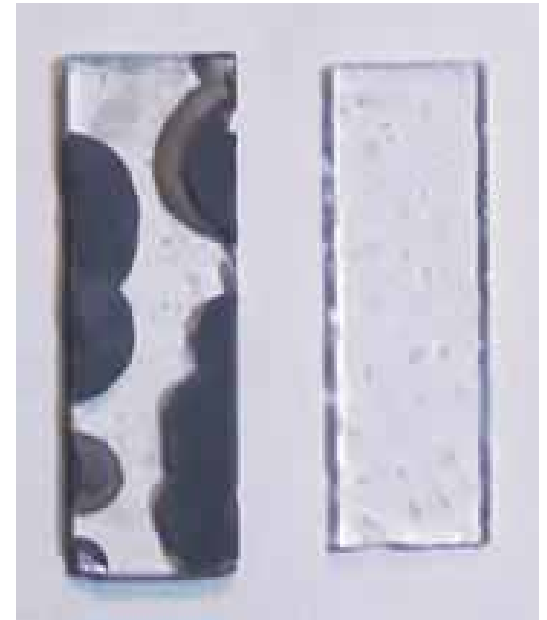
Si Crystal from SiO₂ upon Electron Excitation

- Electron injection through wires causes local reduction of SiO₂, generating Si crystal

[Nohira, Yasuda & Itoh: Nature Materials 2, 397, (2003)]

- Femto-second laser pulses seem to generate Si crystal

[Sokolowski-Tinten et al: PRL 87, 225701 (2001); K. Hirao et al: Jpn APS Meeting; K. Murakami: unpublished]



New technique to grow materials?
If yes, then how is it made? →

*Free Energy MD
Calculation with
Excited electrons*

Electron Excitation

Electron Temperature Increase

- Laser pulses of different frequencies may be simulated via finite electronic temperature molecular dynamics within the Free Energy functional formalism
- This is supposed to reproduce what occurs when SiO_2 is perturbed as in femto-second laser experiments.
- 3 different electronic temperatures are considered, corresponding to 3 different laser frequencies:

$$T[e] = 20000 \text{ K} \quad \longleftrightarrow \quad \hbar\omega = 1.72 \text{ eV}$$

$$T[e] = 25000 \text{ K} \quad \longleftrightarrow \quad \hbar\omega = 2.16 \text{ eV}$$

$$T[e] = 30000 \text{ K} \quad \longleftrightarrow \quad \hbar\omega = 2.59 \text{ eV}$$

Free Energy Molecular Dynamics

Free Energy is written as

$$F[\rho(r), \{R_I\}] = -2k_B T_e \ln \det \left[1 + \exp\left(-\frac{H - \mu}{k_B T_e}\right) \right] + \mu N_e + E_{II} \\ - \int dr \rho(r) \left(\frac{V_H(r)}{2} + \frac{\delta E_{XC}}{\delta \rho(r)} \right) + E_{XC}$$

Then, equations of motion are introduced by the following forces:

$$\frac{\delta F}{\delta \langle \psi_i |} = \frac{\delta F}{\delta \rho(r)} |\psi_i \rangle \quad \text{and} \quad \mathbf{f}_I = \nabla_I F$$

Here the electron charge density is given by

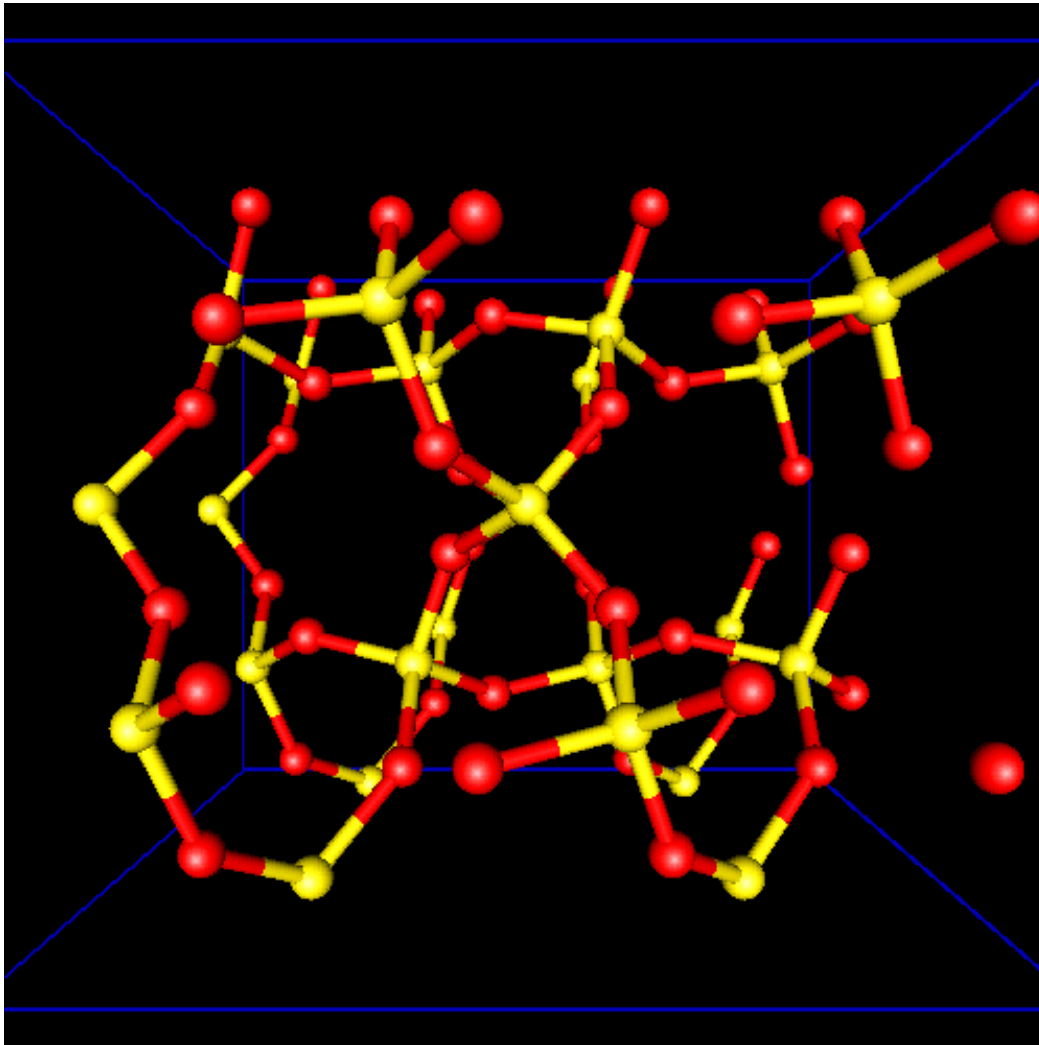
$$\rho(r) = \sum_i n_i |\psi_i(r)|^2$$

[Alavi et al: PRL 73, 2599 (1994)]

with

$$n_i = \frac{1}{1 + \exp[(\epsilon_i - \mu)/k_B T_e]}$$

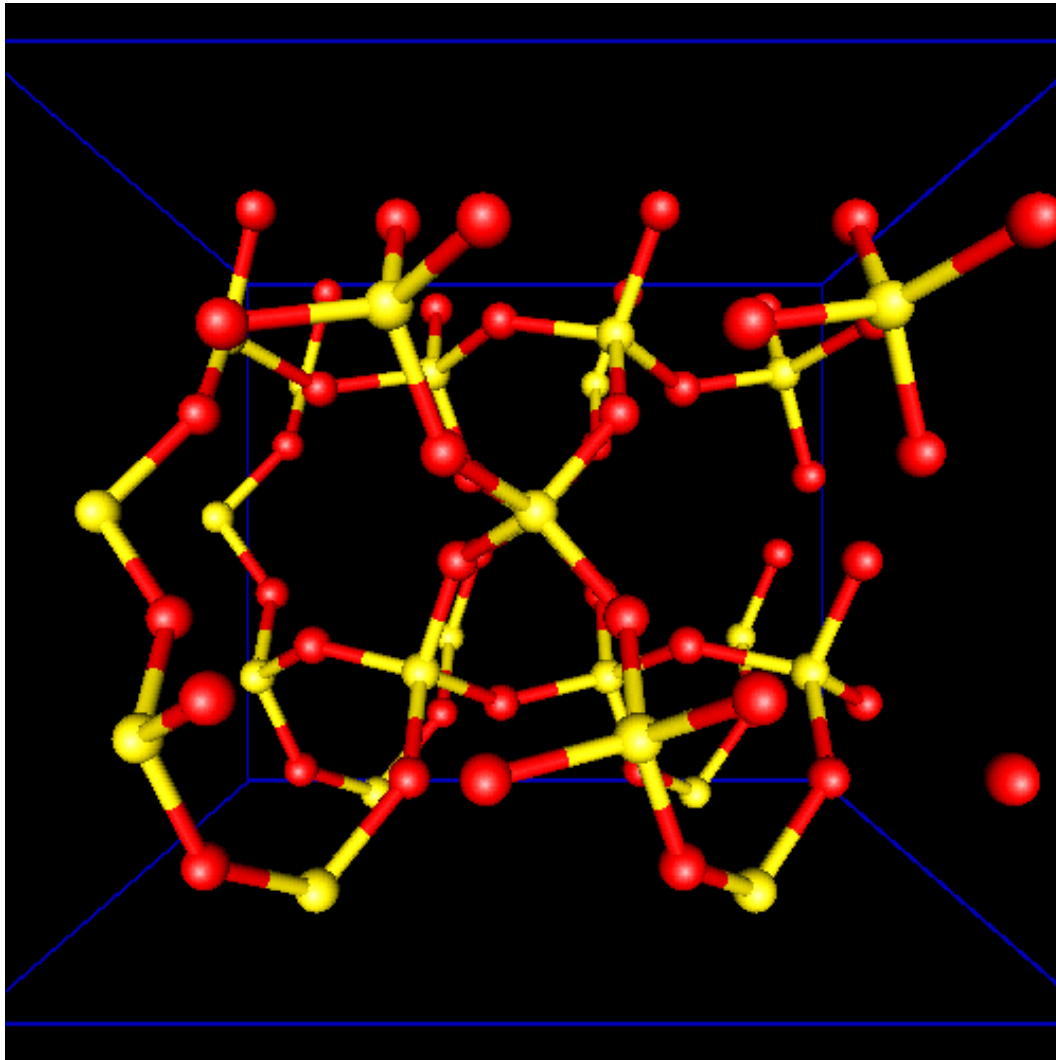
Simulation at $T_e=20000$ K (1.72 eV)



$$\langle T_{\text{ion}} \rangle = 304 \text{ K}$$

Just
a thermal motion

Simulation at $T_e=25000$ K (2.16 eV)

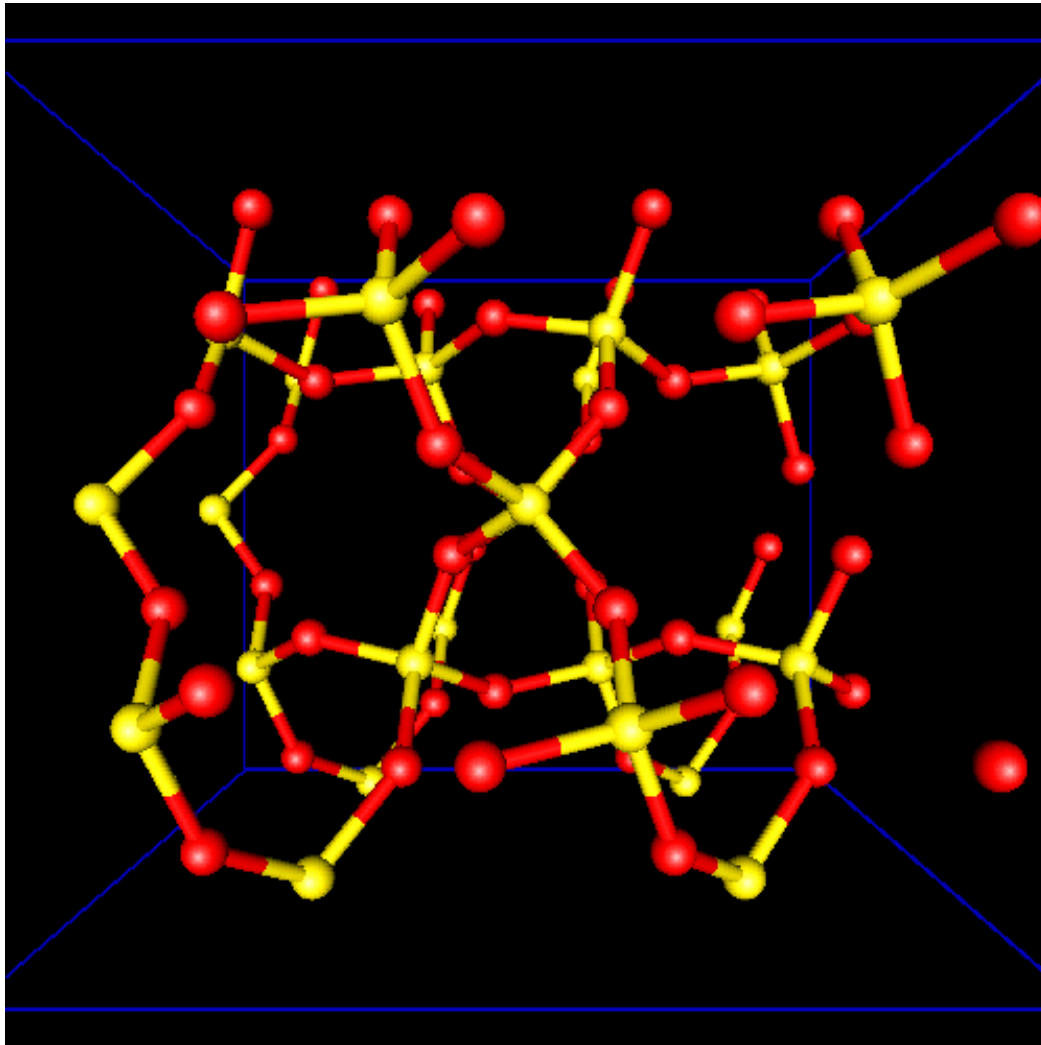


$$\langle T_{\text{ion}} \rangle = 367 \text{ K}$$

BUT!

Formation of
Si-Si bonds

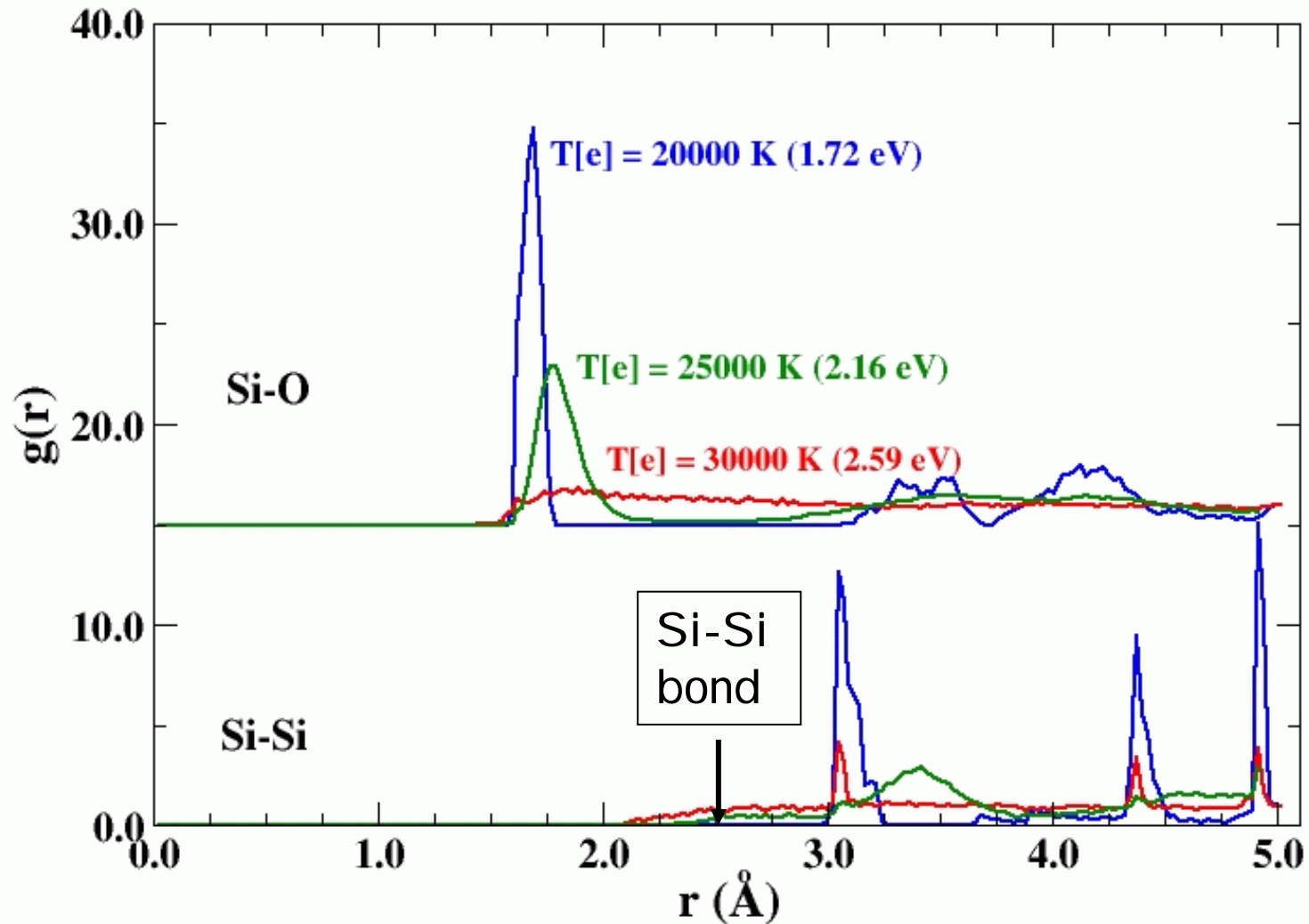
Simulation at $T_e=30000$ K (2.59 eV)



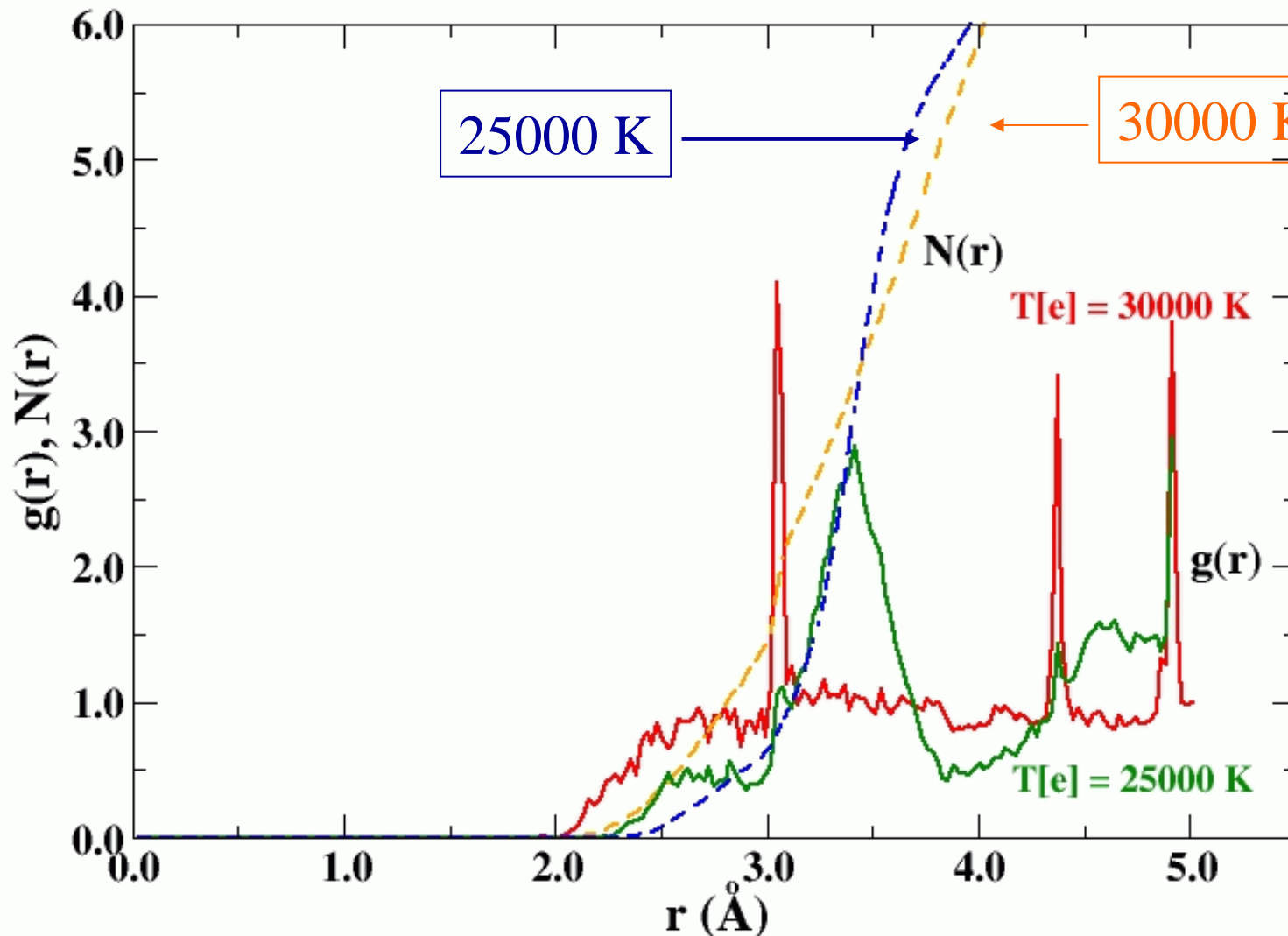
$$\langle T_{\text{ion}} \rangle = 2870 \text{ K} \\ (T_{\text{melt}} = 1883 \text{ K})$$

Just melt

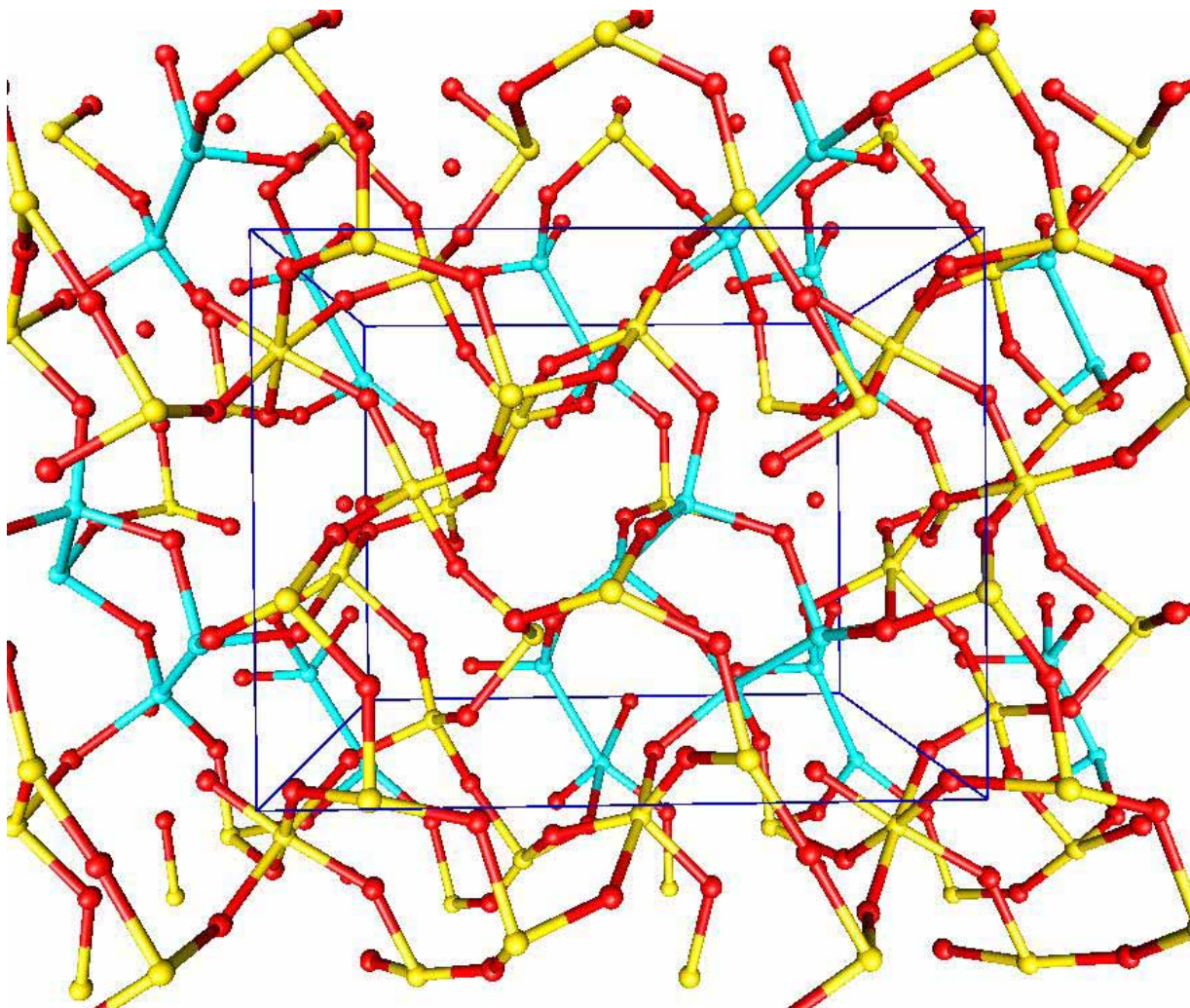
Radial distribution functions for the three electron temperatures



Si-Si RDF and integrated coordination number for the two **electron** temperatures

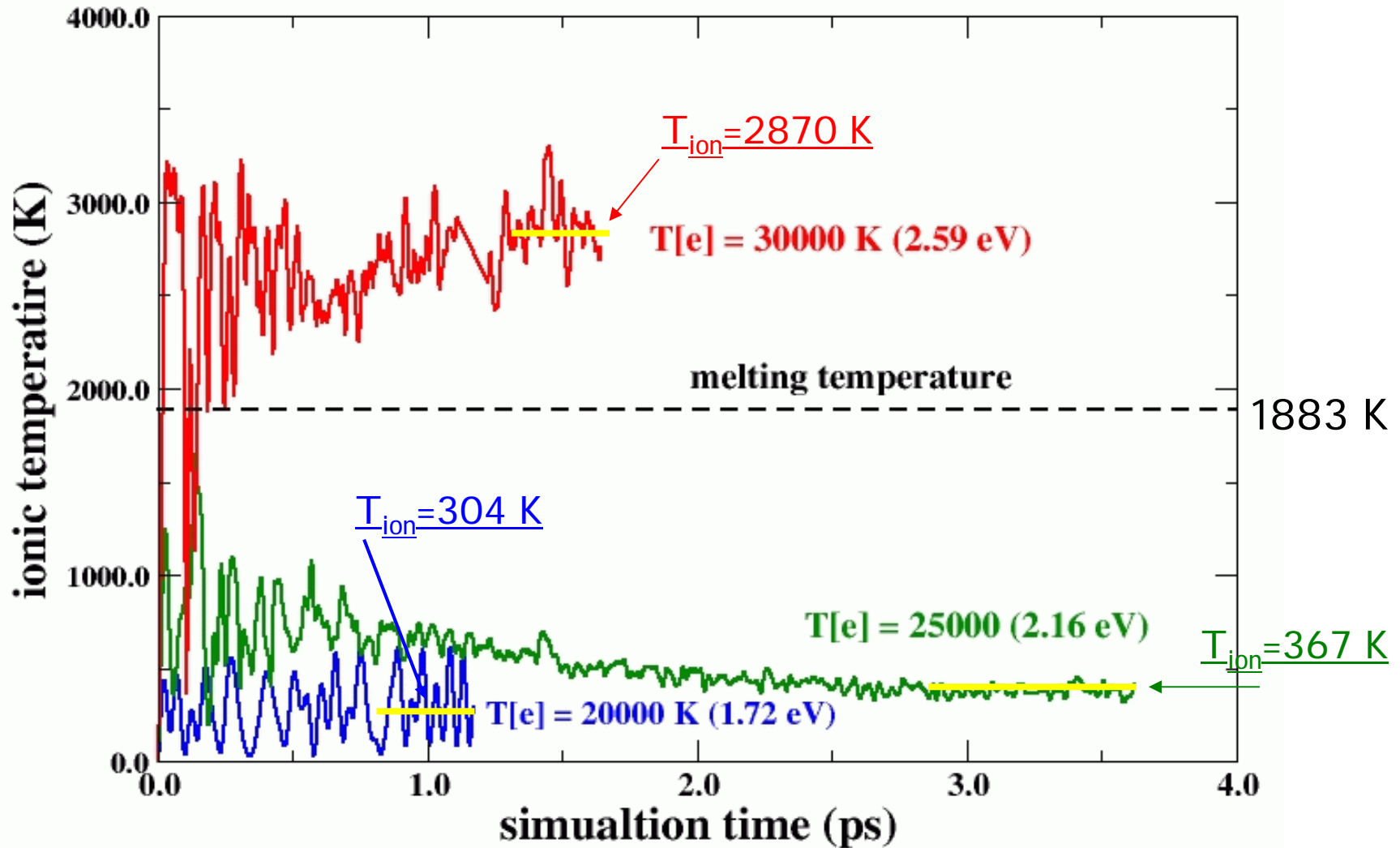


Final stable structure obtained at $T_e = 25000$ K (Si-Si bonds in light blue)



Si-Si bonds
($d=2.51\text{\AA} - 2.64\text{\AA}$)
are 20% of the
total Si related
bonds

Ionic Temperature: Si Formation below Melting Temperature



Diffusion Coefficients for Si & O

T[e] (K)	T_{ion} (K)	D_{Si} (cm ² /s)	D_{O} (cm ² /s)	Si-Si bond
20000	304+/-151	1.004×10^{-9}	1.252×10^{-9}	No
25000	367+/-102	4.100×10^{-9}	8.489×10^{-9}	Yes
30000	2870+/-220	1.415×10^{-7}	3.401×10^{-7}	Yes (but melt)

$$D_j = \frac{1}{3} \int_0^{\Delta t} \left\langle \frac{1}{N_j} \sum_{I=1}^{N_j} \vec{v}_I(t) \vec{v}_I(0) \right\rangle$$

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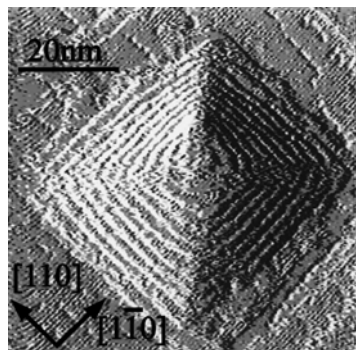
10,000 - 100,000 原子群の量子論的計算

➤ なぜ？

- ナノ・バイオ物質は 10^4 原子群から成る
- 局所的な量子論的化学反应が大局的な構造形成・変化を引き起こし、物としての機能が発現

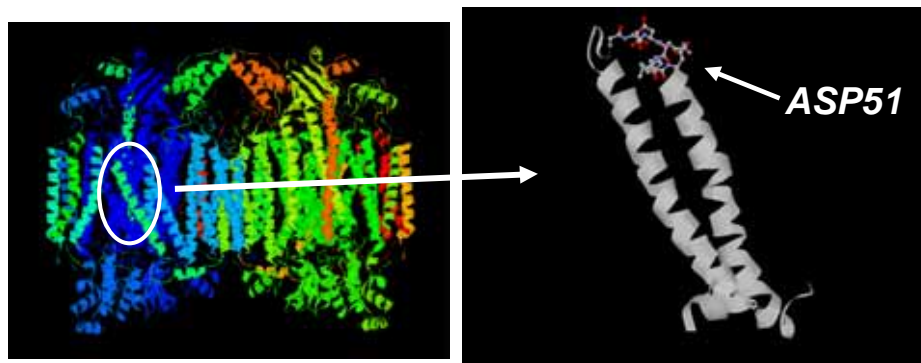
マルチスケールの量子論，形と機能の量子論

原子構造，電子状態，反応機構，...



10nmのSi立方体には5万個の原子

シリコン・ナノピラミッド



シトクローム酸化酵素

10,000 - 100,000 原子群の量子論的計算

➤ どうやって？

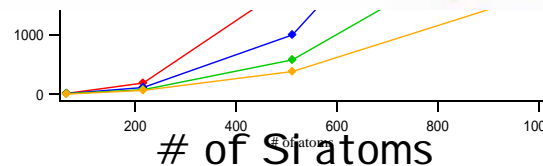
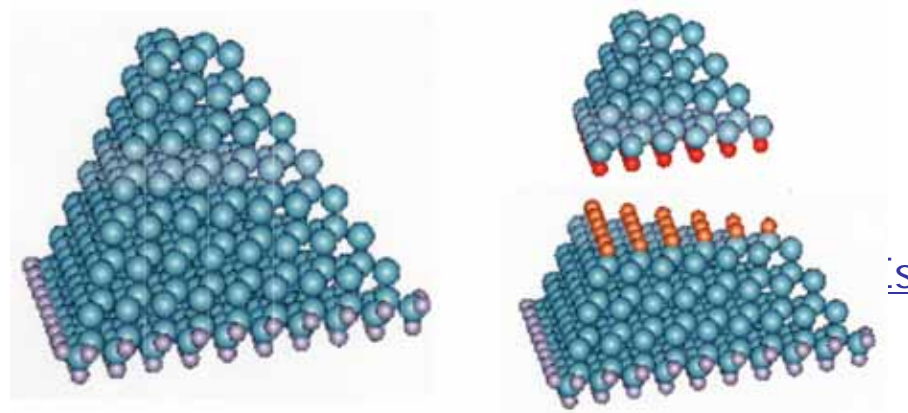
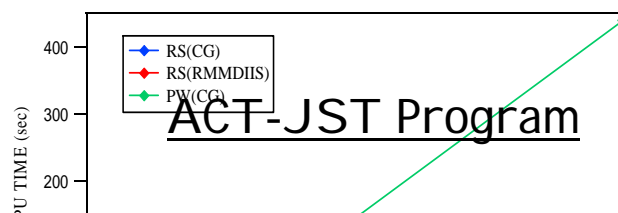
● 10,000 原子群の 密度汎関数法計算

擬ポテンシャル・実空間
差分法をPACS-
CS上で

● 100,000原子群の ハイブリッド計算

DFT-Tight
Binding-MM ハイ
ブリッド

- ✓ 擬ポテンシャルの局所性を活用
- ✓ Residual Minimization
- ✓ No FFT



PACS-CS における物質・生命科学 I: ナノ形状の量子論

- Curvature-induced metallization of semiconducting double-wall carbon-nanotubes
[Okada & Oshiyama: PRL 91, 216801 (2003), and unpublished results.]
 - *Balance of 2 nano-curvatures makes it. The similar happens at metal contacts.*
- C₆₀ badminton shuttlecock: possible magnetism
[Okada et al: CPL 399, 157 (2004).]
 - *Attachment of molecules becomes nano-scissors.*
- Formation of Si seeds in SiO₂ upon laser irradiation
[Boero, Silvestrelli & Oshiyama: 2005.]
 - *Higher electron temperature that simulates electron-excitation-induced materials formation*
- What we are planning to do on PACS-CS